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The synthesis of O-(5-ethoxy-1-methyl-6-oxo-1*H*-pyridazin-4-yl), O-(5-methoxy-1-benzyl-6-oxo-1*H*-pyridazin-4-yl), O-(5-methylthio-1-methyl-6-oxo-1*H*-pyridazin-4-yl) and O-(5-methoxy-1-cyclohexyl-6-oxo-1*H*-pyridazin-4-yl) esters of phosphoric, thiophosphoric and thiophosphonic acids is described. The IR and UV spectra of the synthetized compounds were measured and interpreted and their contact and systemic insecticidal, acaricidal and ovicidal activity was determined. Several of the compounds tested displayed high activity.

In continuation of our studies concerning the synthesis and pesticidal activity of pyridazin-4-yl-esters of organophosphoric acids we concentrated on the synthesis, studies of the spectral data, and the determination of contact and systemic insecticidal, acaricidal and ovicidal activity of pyridazin-4-yl esters of phosphoric, thiophosphoric and thiophosphonic acids of formula *I*:



For this study such groups of substituents on phosphorus were selected which proved most effective in earlier studies concerning pesticidal activity: diethylphosphoryl-, O,O-dimethylthiophosphoryl-, O-methyl-O-ethylthiophosphoryl-, O,O-diethylthiophosphoryl-, O-ethyl-O-isopropylthiophosphoryl-, O-methyl-N-isopropylamidothiophosphoryl- and O-ethyl-ethylthiophosphoryl. On the other hand such substituents were selected for the pyridazine ring which were expected to be very interesting from the point of view of pesticidal activity.

The synthesis of compounds was carried out using the reaction of the chlorophosphoric, chlorothiophosphoric, or chlorothiophosphonic acid esters with the potassium salt of 1,5-disubstituted 6-oxo-1*H*-pyridazin-4-ol in acetonitrile:



SCHEME 1

The esters of chlorophosphoric acid reacted more rapidly, but the products II, IX, XVI and XXIII thus obtained were less stable at room temperature than the corresponding thiophosphates. The stability of thiophosphates was dependent on the substituents R^1 and R^2 and it increased directly with the size of the alkyl. The purity

No	R^1	R ²	х	R ³	R ⁴	
I	CH ₃	CH ₃ O	s	C ₂ H ₅ O	CH ₃	
II	C_2H_5	C ₂ H ₅ O	0	C ₂ H ₅ O	CH ₃	
III	CH ₃	C ₂ H ₅ O	S	C ₂ H ₅ O	CH ₃	
IV	C_2H_5	C ₂ H ₅ O	S	C ₂ H ₅ O	CH ₃	
V	C_2H_5	(CH ₃) ₂ CHO	S	C ₂ H ₅ O	CH ₃	
VI	C_2H_5	C ₂ H ₅	S	C ₂ H ₅ O	CH ₃	
VII	CH ₃	(CH ₃) ₂ CHNH	S	C ₂ H ₅ O	CH ₃	
VIII	CH ₃	CH ₃ O	S	CH ₃ S	CH ₃	1204
IX	C_2H_5	C ₂ H ₅ O	0	CH ₃ S	CH ₃	
Х	C ₂ H ₅	C ₂ H ₅ O	S	CH ₃ S	CH ₃	
XI	CH ₃	C ₂ H ₅ O	S	CH ₃ S	CH ₃	
XII	C_2H_5	(CH ₃) ₂ CHO	S	CH ₃ S	CH ₃	
XIII	C ₂ H ₅	C ₂ H ₅	S	CH ₃ S	CH ₃	
XIV	CH ₃	(CH ₃) ₂ CHNH	S	CH ₃ O	CH ₃	
XV	CH ₃	CH ₃ O	S	CH ₃ O	$C_6 H_{11}$	
XVI	C_2H_5	C ₂ H ₅ O	0	CH3O	C ₆ H ₁₁	
XVII	CH ₃	C ₂ H ₅ O	S	CH ₃ O	$C_{6}H_{11}$	
XVIII	C_2H_5	C ₂ H ₅ O	S	CH ₃ O	C ₆ H ₁₁	
XIX	C_2H_5	(CH ₃) ₂ CHO	S	CH ₃ O	C ₆ H ₁₁	
XX	C_2H_5	C ₂ H ₅	S	CH ₃ O	$C_{6}H_{11}$	
XXI	CH ₃	(CH ₃) ₂ CHNH	S	CH ₃ O	$C_{6}H_{11}$	
XXII	CH ₃	CH ₃ O	S	CH ₃ O	CH ₂ C ₆ H ₅	
XXIII	C_2H_5	C ₂ H ₅ O	0	CH ₃ O	CH ₂ C ₆ H ₅	
XXIV	C ₂ H ₅	C ₂ H ₅ O	S	CH ₃ O	CH ₂ C ₆ H ₅	
XXV	CH ₃	C ₂ H ₅ O	S	CH ₃ O	CH ₂ C ₆ H ₅	
XXVI	C_2H_5	(CH ₃) ₂ CHO	s	CH ₃ O	CH ₂ C ₆ H ₅	
XXVII	C ₂ H,	C_2H_5	S	CH ₃ O	CH ₂ C ₆ H ₅	
XXVIII	СĤ3	(CH ₃) ₂ CHNH	S	СН₃О	CH ₂ C ₆ H ₅	

Esters of Organophosphoric Acids

TABLE I

Compounds Prepared

	Composition	Calculate	d/Found	n ²⁰
Compound	(M.w.)	% P	% S	(yield, %)
Ι	C ₉ H ₁₅ N ₂ O ₅ PS	10·53	10·90	1·5311
	(294·3)	10·50	11·16	(68)
II	$C_{11}H_{19}N_2O_6P_{(306\cdot2)}$	10·11 9·82	-	1·4869 (91)
111	$C_{10}H_{17}N_2O_5PS$	10·05	10·40	1·5249
	(308·3)	10·03	10·62	(84)
IV	$C_{11}H_{19}N_2O_5PS$	9·61	9·95	1·5183
	(322.3)	9·98	10·15	(87)
ν	$C_{12}H_{21}N_2O_5PS$	9·21	9·53	1·5120
	(336.4)	9·13	9·71	(91)
VI	$C_{11}H_{19}N_2O_4PS$	10-11	10·47	1·5411
	(306.3)	10-22	10·60	(79)
VII	$C_{11}H_{20}N_{3}O_{4}PS$	9∙69	9·98	1·5371
	(321.3)	9∙84	10·05	(78)
VIII	C ₈ H ₁₃ N ₂ O ₄ PS ₂	10·45	21·64	1·5592
	(296·3)	10·38	21·82	(67)
IX	C ₁₀ H ₁₇ N ₂ O ₅ PS	10∙05	10·40	1·5170
	(308·4)	9∙72	10·22	(72)
X	$C_{10}H_{17}N_2O_4PS_2$	9·55	19·77	1·5540
	(324·3)	9·36	19·55	(84)
XI	C ₉ H ₁₅ N ₂ O ₄ PS ₂	9·98	20·66	1·5634
	(310·3)	10·11	20·88	(88)
XII	$C_{11}H_{19}N_2O_4PS_2$	9·15	18-95	1·5498
	(338·4)	8·93	18-61	(77)
XIII	$C_{10}H_{17}N_2O_3PS_2$	10∙05	20·80	1·5815
	(308.3)	9∙91	20·28	(89)
XIV	$C_{10}H_{18}N_3O_3PS_2$	9·58	19·83	1·5866
	(323·3)	9·32	20·04	(80)
XV	C ₁₃ H ₂₁ N ₂ O ₅ PS	8·89	9·20	1·5399
	(348·3)	9·67	9·45	(86)
XVI	C ₁₅ H ₂₅ N ₂ O ₆ P (360·3)	8·59 8·51	_	1·5055 (78)
XVII	C ₁₄ H ₂₃ N ₂ O ₅ PS	8-55	8-85	1·5342
	(362·4)	8-44	9-02	(85)

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TABLE I

(Continued)

C	Composition	Calculate	n ²⁰	
Compound	(M.w.)	%Р	% S	(yield, %)
XVIII	C ₁₅ H ₂₅ N ₂ O ₅ PS	8-23	8-52	1·5280
	(376·4)	8-18	8-47	(96)
XIX	C ₁₆ H ₂₇ N ₂ O ₅ PS	7·93	8·21	1·5227
	(390·5)	7·94	8·40	(91)
XX	C ₁₅ H ₂₅ N ₂ O ₄ PS	8·59	8·90	1·5490
	(360·4)	8·49	8·63	(69)
XXI	C ₁₅ H ₂₆ N ₃ O ₄ PS	8·25	8∙54	<i>a</i>
	(375·4)	8·34	8∙70	(64)
XXII	C ₁₄ H ₁₇ N ₂ O ₅ PS	8.69	9·00	1·5709
	(356·3)	8.66	8·78	(58)
XXIII	C ₁₆ H ₂₁ N ₂ O ₆ P (368·3)	8·41 8·50		1·5325 (64)
XXIV	C ₁₆ H ₂₁ N ₂ O ₅ PS	8·06	8·34	1·5516
	(384·4)	8·12	8·52	(72)
XXV	C ₁₅ H ₁₉ N ₂ O ₅ PS	8·36	8·66	1·5672
	(370·3)	8·40	8·88	(73)
XXVI	C ₁₇ H ₂₃ N ₂ O ₅ PS	7·77	8·05	1·5484
	(398·4)	7·91	8·19	(84)
XXVII	$C_{16}H_{21}N_2O_4PS$	8·41	8·70	1·5728 ^(**)
	(368.4)	8·35	8·90	(71)
XXVIII	$C_{16}H_{22}N_{3}O_{4}PS$	8·08 7·92	8·36 8·48	1.5836

^a M.p. 72-73°C.

of the phosphates prepared was very high; however, some thiophosphates had to be purified by column chromatography. The purity of the substances was followed by thin-layer chromatography (Table I).

In the IR spectra of the synthetized compounds a very intensive band of the v(C=O) stretching vibration is observed at 1648-1658 cm⁻¹, then a medium strong band v(C=N) in compounds I-VII, XV-XXVIII at 1618-1620 cm⁻¹, while in compounds VIII-XIV it was at 1578-1582 cm⁻¹, further a weak band of v(C=C) in compounds I-VII, XV-XVIII at 1531-1540 cm⁻¹ or in compounds VIII-XIV at 1602-1609 cm⁻¹. The moderately intensive band at 1370-1388 is

due to the bending vibration $\delta(=CH-)$. The stretching vibration v(C-N) at 1260 to 1279 is of medium intensity. In compounds containing the P=S bond the first stretching vibration band v(P=S) is at $668-675 \text{ cm}^{-1}$ for compounds VIII-XIV while for other compounds it is at $660-667 \text{ cm}^{-1}$. The second band of compounds I-XXI is at $685-720 \text{ cm}^{-1}$, while it is absent in the case of compounds XXII to XXVII. For compounds VIII-XIV the stretching vibration band of the v(P=O) stretching vibration is at $1280-1300 \text{ cm}^{-1}$. For compounds VII, XIV and XXVII the intensive band of the v(P=O) stretching vibration is at $1280-1300 \text{ cm}^{-1}$. For compounds VII, XIV and XXVIII the stretching vibration band v(N-H) occuring at $3408-3412 \text{ cm}^{-1}$ is characteristic, while the second band at 3262 to 3280 cm^{-1} can be assigned to the bond formed by the amide group hydrogen and the sulphur atom bound to phosphorus (Table II).

In the ultraviolet spectra of the synthetized compounds two to three maxima are present. The first maximum is in all cases at 211-215 nm, in the case of compounds VIII-XIV the second maximum is at 237-238 nm and the third at 320-321 nm. In other compounds only the second maximum is present, at $285\cdot5-290$ nm. The bands at the shortest wave-lengths belong to $\pi \rightarrow \pi^*$ transitions, while the bands at the longest wave-lengths belong to $n \rightarrow \pi^*$ transitions (Table II).

During the testing of the insecticidal activity none of the substances was so active in the first screening as to warrant a further more accurate testing. In the test for contact insecticidal activity on Musca domestica the substances studied were very well active, with a few exceptions (V, IX - XII, XV, XXII and XXV). More than a half were more active than the standard malathion used, while substances I, VI and IX were about as active as phenitrothion; in the test on Calandra granaria generally less substances were active. Thus, in comparison with malathion only compound IX was more active, while in the test on Aphis fabae the highest activity could be measured. All the substances were submitted to more accurate tests. More active than both standards used (malathion and phenitrothion) were compounds III, V, VI, VII and XII, while substances IV, XI, XIV and XXII were about equally active. For the test with M. domestica carried out with respect to time 12 substances were selected for more accurate tests. All but two of them were more active than malathion.



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TABLE II

Spectral Data of the Synthetized Compounds

(Com-	I-	IR Spectra, cm ⁻¹					
р	ound $\nu(C=0)$	ν(C==) ν(C=N)	ν(C==C)	δ(==CH) v(C—N)	$\nu(P=S)$	
I	1 651	1 65	1 619	1 534	1 380	1 261	662 686	
II	1 653	1 65	1 620	1 533	1 380	1 263		
11.	1 655	1 65	1 618	1 535	1 380	1260	663 687	
IV	1 654	1 65	1 619	1 535	1 389	1 261	661 686	
V	1 656	1 65	1 619	1 536	1 385	1 260	662 686	
V	1 655	1 65	1 620	1 534	1 381	1 260	662 685	
V	1 646	1 64	1 615	1 531	1 380	1 262	661 684	
V	1 651	1 65	1 582	1 605	1 375	1 267	674 705	
13	1 652	1 65	1 581	1 606	1 374	1 268	-	
X	1 650	1 65	1 581	1 605	1 373	1 267	675 704	
X	1 650	1 65	1 581	1 606	1 373	1 265	-673 706	
X	1 651	1 65	1 582	1 609	1 374	1 374	672 701	
X	1 652	1 65	1 578	1 606	1 378	1 265	670 703	
X	IV 1 648	1 64	1 578	1 602	1 376	1 263	668 702	
X	V 1 650	1 65	1 620	1 540	1 373	1 278	665 710	
X	VI 1648	1 64	1 620	1 537	1 370	1 279	_	
X	VII 1 650	1 65	1 620	1 540	1 371	1 273	663 706	
X	<i>VIII</i> 1 650	1 65	1 619	1 538	1 370	1 276	663 708	
XI	X 1 648	1 64	1 620	1 540	1 374	1 278	662 707	

TABLE II

(Continued)

Various		UV Spectra	λ_{\max} nm	(log ε)	
		212.5 (4.31)	285.5 (3.72)		
ν(P==Ο)	1 280	212.0 (4.31)	286.0 (3.73)		
		212.0 (4.28)	287.0 (3.70)		
		212.5 (4.28)	286.0 (3.70)		
		211.5 (4.35)	287.0 (3.73)		
		213.0 (4.32)	288-0 (3-68)		
ν(NH)	3 412 3 268	212.5 (4.43)	287.5 (3.66)		
ν(CH ₃ SC)	648	211.0 (4.22)	237.5 (3.70)	320-0 (3-86)	
$\nu(CH_3 - S - C)$ $\nu(P = O)$	647 1 289	212.0 (4.18)	238-0 (3-66)	321.0 (3.81)	
v(CH ₃ -S-C)	647	211.5 (4.20)	238-0 (3-72)	320.0 (3.82)	
v(CH ₃ —S—C)	650	211.0 (4.13)	237.0 (3.75)	319.0 (3.84)	
v(CH ₃ —S—C)	648	210.0 (4.19)	237.0 (3.76)	320.0 (3.80)	
v(CH ₃ —S—C)	645	212.0 (4.20)	238.0 (3.72)	320.0 (3.79)	
ν(CH ₃ —S—C) ν(NH)	643 3 410	212.5 (4.19)	237-0 (3-74)	321.0 (3.80)	
	3 270	214.0 (4.32)	288.0 (3.74		
(P==0)	1 292	213.0 (4.32)	289.0 (3.74)		
		214.0 (4.31)	288.0 (3.73)		
		213-5 (4-27)	288-5 (3-68)		
		215.0 (4.24)	283.0 (3.72)		

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Table 11

(Continued)

Com-	IR Spectra, cm ⁻¹						
pound	v(C==0)	v(C==N)	ν(C==C)	$\delta = CH$	-) v(CN)	v(P==S)	
XX	1 650	1 618	1 535	1 372	1 277	660 697	
XXI	1 658	1 612	1 536	1 387	1 272	660 700	
XXII	1 656	1 620	1 532	1 387	1 278	661	
XXIII	1 656	1 620	1 532	1 387	1 278	-	
XXIV	1 858	1 618	1 534	1 386	1 272	667	
XXV	1 657	1 619	1 532	1 386	1 271	665	
XXVI	1 658	1 620	1 534	1 388	1 271	664	
XXVII	1 658	1 618	1 533	1 387	1 270	667	
XXVIII	1 652	1 612	1 538	1 388	1 272	661 694	

Compounds I-IV, VI, XVI-XVIII and XXIII were also highly active which exceeded the activity of malathion 2-3 times. Except for substance VII a very good activity was also found in the test on acaricidal activity tested on the females of *Tetranychus urticae*. Remarkably high activity was also found in compound XXVI, *i.e.* a double with respect to the standard carbophenothion, while the activity of compounds V, XX and XXIV is also remarkable. In comparison with Acrex as standard 18 compounds were more active, those most active almost by two orders of magnitude. In the test on the eggs of T: *urticae* a remarkable activity was found for compounds XXIV, which were about equally active as the standards used.

From the total estimation of the insecticidal, acaricidal and ovicidal activity it follows that the best results were obtained in the tests on contact insecticidal and acaricidal activity. It is impossible to decide unambiguously which substituent $(R^1 - R^4)$ affects the pesticidal activity and how, even though it seems that in contact insecticidal activity on *A. fabae*, acaricidal and ovicidal activity on *T. urticae*, O-ethyl-O-isopropylthiophosphoryl derivatives of the following formula were the most active:

.1

TABLE II

(Continued)

Var	Various		λ _{max} nm	(log ε)
		215.0 (4.38)	288-5 (3-77)	
v(NH)	3 411 3 262 ,	213.5 (4.30)	287-5 (3-72)	
		213.0 (4.37)	288-0 (3-80)	
v(P==O)	1 300	212.5 (4-40)	287.0 (3.74)	
		212.5 (4.39)	287.5 (3.72)	
		213.0 (4.38)	288-0 (3-80)	
		212.0 (4.34)	288.5 (3.67)	
		214.0 (4.43)	288-5 (3-74)	
v(NH)	3 408 3 280	213.0 (4.48)	287.0 (3.62)	



EXPERIMENTAL

Methods

The 1R spectra (400-2200 cm⁻¹) were measured on a Zeiss IR Specord 75 instrument. The wave-number calibration was carried out using the spectrum of a polystyrene foil. The spectra were measured in CCl₄ in a 0·1 mm cell (concentration 10⁻²). The UV spectra were measured on a Unicam SP 8000 spectrophotometer in methanol. The calibration was carried out with a Holmi filter. The concentration of the solutions measured was 2 $\cdot 10^{-5} - 5 \cdot 5 \cdot 10^{-5}$ M, in 1 cm cells. Thin-layer chromatography was carried out on aluminum foils with a layer of silica gel (Silufol "R" without indicator from Lachema, Brno) in benzene-acetone (9:1->8:2). Detection with 0.5% 2,6-dibromoquinone-4-chloroimide in light petroleum, at 120°C. Column chromato-

graphy was carried out on Silica gel L 100/160 mesh for column chromatography (Lachema, Brno). Before use the silica gel was activated at 140°C for 6 h. Toluene with an addition of acetone (from 0 to 5%, depending on the character of the impurities) was used for elution. The course of the separation was followed by means of TLC.

TABLE III

Insecticidal, Acaricidal and Ovicidal Activity of the Synthetized Compounds

		Time effect				
Compound	M domestica	C anonania	A fahaa	T. urt	icae	LT ₅₀ , min
	M. aomestica	C. granaria	A. Jabae	females	eggs	M. domestica
I	0.0020	>0.1	0.0067	0.00012	0.016	6.2
II	0.0091	>0.1	0.0042	0.00027	0.022	7.0
111	0.0098	>0.1	0.00020	0.00020	0.038	7.0
IV	0.0105	0.021	0.0012	0.00017	0.017	5.0
V	0.10	0.0104	0.000123	0.0000817	0.012	_
VI	0.0046	>0.1	0.00078	0.00028	0.042	7-5
VII	0.0102	>0.1	0.00098	>0.1	>0.2	12.0
VIII	0.45	>0.1	0.0031	0.00048	>0.2	
IX	0.0062	0.0021	0.0032	0.0011	>0.2	·
Х	>0.2	>0.1	0.0019	0.00092	0.11	
XI	>0.2	>0.1	0.0011	0.00019	0.10	_
XII	>0.2	>0.1	0.000958	0.00037	>0.2	
XIII	0.0081	0.049	0.0030	0.00057	0.48	
XIV	0.048	0.061	0.0019	0.00108	>0.2	_
XV	>0.2	0.092	0.0030	0.0041	0.10	
XVI	0.029	0.10	0.010	0.00074	>0.2	6:5
XVII	0.023	0.011	0.011	0.00021	0.014	8·5 ^ˆ
XVIII	0.016	0.048	0.0057	0.00017	0.021	7.5
XIX	>0.2	>0.1	0.020	0.000202	0.1	
XX	0.050	>0.1	0.018	0.000070	>0.2	18.0
XXI	0.0236	>0.1	0.088	0.00140	0.44	20 0
XXII	>0.2	>0.1	0.0020	0.082	0.11	
XXIII	0.0102	>0.1	0.0076	0.041	0.016	4.0
XXIV	0.32	0.082	0.0042	0.000075	0.0043	
XXV	>0.2	>0.1	0.0053	0.000103	0.000103	-
XXVI	0.088	0.079	0.070	0.0000175	0.0031	
XXVII	0.10	>0.1	0.016	0.000103	0.0020	
XXVIII	0.12	>0.1	0.0022	0.055	>0.2	_
Malathion	0.036	0.0076	0.0011	0.00241	_	13.0
Phenitro- thion	0.0040	0.00069	0.0010	0.012	-	_
Carbopheno thion	>0-1	>0.1	0.00064	0.000037	0.0026	
Acrex	_	-	-	0.0023	0.0043	-

Synthesis of Substances

5-Ethoxy-1-methyl-6-oxo-1*H*-pyridazin-4-ol with m.p. 140-142°C, 5-methylthio-1-methyl-6oxo-1*H*-pyridazin-4-ol with m.p. 188-190°C, 1-cyclohexyl-5-methoxy-6-oxo-1*H*-pyridazin-4-ol with m.p. 163-165°C, and 1-benzyl-5-methoxy-6-oxo-1*H*-pyridazin-4-ol with m.p. 147-149°C were prepared according to ref.¹.

Compounds II, IX, XVI and XXIII: Diethyl ester of chlorophosphoric acid (0.05 mol) was added to 0.055 mol of the potassium salt of 1,5-disubstituted 6-oxo-1*H*-pyridazin-4-ol in 100 ml acetonitrile at 10°C and under stirring. The stirring was continued at 40°C for 4h. After cooling the separated salt was filtered off and acetonitrile was evaporated from the filtrate under reduced pressure. The residue was dissolved in 100 ml of toluene, washed with water, 5% sodium carbonate solution, and again with water. After drying the solution toluene was distilled off in a vacuum.

Compounds I, III-VIII, X-XV, XVII-XXII, XXIV-XXVIII: Ester of chlorothiophosphoric acid or chlorothiophosphonic acid (0.055 mol) was added to 0.06 mol of the potassium salt of 1,5-disubstituted 6-oxo-1H-pyridazin-4-ol in 100 ml of acetonitrile and the reaction mixture was stirred and refluxed for 2 to 4 h. After cooling 100 ml of toluene were added, the solution was washed with water, 5% sodium carbonate solution and water. After drying over sodium sulfate toluene was distilled off under reduced pressure. The residue was purified by column chromatography as necessary.

Pesticidal Activity

Contact insesticidal activity was tested on domestic fly (Musca domestica L.), Calandra granaria L., and Aphis fabae ScoP., using malathion (O,O-dimethyl-S-1,2-(diethoxycarbonyl)ethyl dithiophosphate) and phenitrothion (O,O-dimethyl-O-(3-methyl-4-nitrophenyl) thiophosphate) as standards. Systemic insecticidal activity was tested on Macrosiphoniela sanborni L. (the host plant was Chrysanthemum indicum) using thiometone (O,O-diethyl-S-(2-ethylthioethyl) dithiophosphate) as standard. The acaricidal activity was assayed on spiders (*Tetranychus urticae* KOCH), the ovicidal activity on the eggs of spiders (*T. urticae*), using carbophenothion (O,O-diethyl-S-(4-chlorophenylthiomethyl) dithiophosphate) and Acrex (2-sec-butyl-4,6-dinitrophenyl isopropyl carbonate) as standard. The methods for the determination of the insecticidal, acaricidal and ovicidal activity have already been published^{2,3}. The insecticidal effect in time was dedermined on M. domestica using malathion as standard, according to a method already described⁴. The results obtained are summarized in Table III.¹

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